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AdSCR systems (NOx Adsorption + Selective Catalytic Reduction) to minimize cold start NOx emissions from Diesel engines

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Graphical Abstract



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Abstract

Most of NOx emissions from Diesel engines occurs during the cold-start period, due to poor activity of SCR converters and to the urea injection temperature threshold [1]. The present work presents a novel multifunctional catalyst able to minimize the release of NOx at low temperature [2-4]. The strategy of the AdSCR system is to add a storage functionality to a state-of-the-art SCR catalyst, allowing both the capture of NOx at room temperature, and the direct reduction of the stored NOx at higher temperatures in a single device. Thanks to its PGM-free formulation, it can replace conventional SCR converters without modifications of the aftertreatment configuration. An experimental investigation was carried out over different samples, prepared by physically mixing a commercial SCR catalyst (Cu-CHA or Fe-ZSM5) and a NOx storage material (BaO/Al₂O₃ or CeO_2/Al_2O_3). First, NO (500 ppm) + O₂ (8%) isothermal adsorption + TPD experiments were performed to estimate the NOx storage capacity of each catalytic system. Then, the dual functionality of such physical mixtures was proved by means of cold start experiments: NO (300 ppm) was fed at room temperature, followed by a T-ramp (15 °C/min); when the temperature reached 170°C (here regarded as a T-threshold for urea injection), NH₃ (800 ppm) was cofed to the reactor to start the SCR reactions. Different runs have been performed to study the effect of the feed composition (presence of H₂O and CO₂) and of preadsorbed ammonia on AdSCR transient behaviour. The results indicated that AdSCR systems are able to adsorb NO in O_2 already at low temperature. In this regard, Cu-CHA + BaO/Al₂O₃ systems showed the best performances. In cold-start runs under dry conditions: (1) after the NO step feed, the NO outlet concentration exhibited a long dead time (420 s) before NOx breakthrough; (2) the overall engine-out NOx were reduced by 93% before NH₃ injection; (3) as soon as NH₃ was fed, a huge peak of N_2 (900 ppm) was released and the NO concentration started to decrease due to the onset of the Standard SCR reaction. Interestingly, additional data showed that the presence of preadsorbed ammonia can even enhance the low temperature SCR activity, resulting in the light-off of nitrogen production already at 80°C. Dedicated tests also showed a negative impact of water and CO₂ on the AdSCR performances; however, such an effect could be counterbalanced by optimizing the ratio between the NOx storage material and the SCR catalyst, and by selecting NOx storage materials more tolerant to CO_2 (such as CeO_2). These promising results may warrant further studies aimed at investigating the new AdSCR concept.

Keywords: Cold start; NOx adsorber; NH₃-SCR, AdSCR.

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Biography of Presenting Author



Isabella Nova is full Professor in Industrial Chemistry at the Department of Energy, Politecnico of Milan, Italy. She has got her Degree in Chemical Engineering (Politecnico di Milano, 1996), her P.h.D.in Industrial Chemistry (1999) from university of Milano. Isabella Nova's research work is focused on catalytic technologies for emission control of lean burn exhausts (Selective Catalytic Reduction of NOx by NH3 and SDPF (SCR+Diesel Particulate Filter), NOx Storage-Reduction, both for mobile and stationary applications), and on new photo-electrochemical processes for the production of clean energy, such as the water splitting, and CO2 conversion in fuels and chemicals. She has co-authored over 150 publications on national and international journals in the fields of Chemical Reaction Engineering and

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